#### [1. Chemical product and company identification ]

Chemical substance name:	Mixture of	Zinc stearate Magnesium stearate
Product name:	Daiwax MZH	
Company name:	Dainichi Chemical	Industry Co., Ltd.
Address:	7-3-4, Nakaishikiri	-cho, Higashiosaka-shi, Osaka-fu,
	579-8014, Japan	
Associated department:	Technical departme	ent
Telephone number:	+81-72-985-1851	
Emergency contact number:	+81-72-985-1851	
FAX number:	+81-72-987-0170	
Recommended use:	Additives for resin	

#### [2. Hazards identification]

1. (	GHS	classification
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a. Physical hazards

Flammable solids:	Class
Pyrophoric solids:	Class
Self-heating substances and mixtures:	Class
Substances and mixture which, in	
contact with water, emit flammable	Class
gases:	
Corrosive to metals:	Class

#### b. Health hazards

Acute toxicity (oral): Acute toxicity (dermal): Acute toxicity (gases): Acute toxicity (vapors): Acute toxicity (vapors): Acute toxicity (dusts and mists): Skin corrosion/irritation: Serious eye damage / eye irritation: Respiratory sensitization: Skin sensitization: Germ cell mutagenicity: Carcinogenicity: Reproductive toxicity: STOT-single exposure: STOT-repeated exposure: Aspiration hazard: Classification not possible Classification not possible Classification not possible

Classification not possible

Classification not possible

Not classified Classification not possible Not applicable Classification not possible Not classified Category 3 Category 2B Classification not possible Not classified Not classified Classification not possible Classification not possible Category 3 Classification not possible Classification not possible

c. Environmental hazards Acute aquatic hazard: Chronic aquatic hazard: Hazardous to the ozone layer:	Classification not possible Classification not possible Not applicable
2. Label elements	
Signal words:	Warning
Hazard pictogram:	!
Hazard statement	•
H316:	Causes mild skin irritation.
H320:	Causes eye irritation.
H335:	May cause respiratory irritation.
Precautionary statement	
Prevention	
P261:	Avoid breathing dust/fume/gas/mist/vapors/spray.
P264:	Wash eyes thoroughly after handling.
P271:	Use only outdoors or in a well-ventilated area.
Response	
P304+P340:	IF INHALED: Remove person to fresh air and keep comfortable for breathing
P305+P351+P338:	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P337+P313:	If eye irritation persists: Get medical advice/attention.
P332+P313:	If skin irritation occurs: Get medical advice/attention.
P312:	Call a doctor or a physicians if you feel unwell.
Storage	r frida i frida
P405:	Store locked up.
P430+P233:	Store in a well-ventilated place. Keep container tightly closed.
Disposal	
P501:	Dispose of contents/container in accordance with local/regional/national/ international regulations.

### [3. Composition/Information on ingredients]

#### Substance/Mixture:

Mixture

Chemical name	Zinc stearate	Magnesium stearate
Compounding ratio (%)	Non-disclosure	Non-disclosure
CAS registry number	557-05-1	557-04-0
ENCS number	(2)-615	(2)-615
EINECS number	209-151-9	209-150-3

#### [4. First-aid measures]

### IN EACH CASES OF FOLLOWING EMERGENCIES, VICTIMS SHOULD BE TREATED BY PARTICULAR FIRST-AID MEASURES AS FOLLOW

In eyes:	Flush eyes with plenty of water for at least 15 minutes. Then get immediate medical advice.
On skin:	Wash skin with soap and water for at least 15 minutes while removing contaminated clothing and shoes. Get medical advice, if needed. Thoroughly clean and dry contaminated clothing and shoes before reuse.
Inhalation:	If adverse effects occur, remove to uncontaminated area. Give artificial respiration if not breathing. If breathing is difficult, oxygen should be administered by qualified personnel. Get immediate medical advice.
Ingestion:	If large amount is swallowed, get medical advice.
[5. Fire-fighting measures ]	
Suitable Extinguishing media:	Water spray, foam-extinguisher, powder-extinguisher and dry chemical
Inappropriate extinguishing media:	Straight stream water

Slight fire hazard.

extinction.

environment.

Dust/air mixtures may ignite or explode.

Take away a product container from a fire if possible. Keep containers cool with a plenty of water after fire

Fire-fighters should wear an appropriate respiratory apparatus and protective clothes for chemical.

Flammable properties:

Special protective actions for fire-fighters:

#### [6. Accidental release measures]

Personal precautions:	Use proper protective equipment as indicated in Section 8. Avoid direct contact with the spilled or leaked material. Avoid inhaling this product in the air (Powder dust).
Methods and materials for containment and cleaning up:	Rake spills with a broom and collect it in appropriate container. Store the container in a cool and dry place until it disposes. Ventilate the area where this product was released.
Environmental precautions:	Avoid flowing out to the rivers, household drains and other

### **[7. Handling and storage ]**

<b>[7. Handling and storage ]</b>	
Handling:	Avoid contact with eyes, skin and clothes. Wash hands carefully after handling this product. Prohibit open flames while handling this product. Avoid deposition of this product. Use dust explosion-proof electrical equipment and light fixtures. Do not eat, drink or smoke while handling this product.
Storage:	Store this product in well-ventilated, dry and cool place. Please make sure that the storage is not close to open flames, sparks and heat. Please make sure that the container of this product is tightly closed when store this product.
[8. Exposure controls/Personal protection ]	
Component Exposure Limit Zinc stearate NIOSH: OSHA (US): Magnesium stearate ACGIH Engineering controls: Ventilation:	<ul> <li>10 mg/m3 TWA total dust; 5 mg/m3 TWA respirable dust</li> <li>15 mg/m3 TWA total dust; 5 mg/m3 TWA respirable dust</li> <li>10 mg/m3 TLV-TWA (Stearates)</li> <li>Eye washer and safety shower should be placed in storages where this product is stored and in buildings where this product is handled.</li> <li>Provide local exhaust ventilation system. Ventilation equipment should be explosion-resistant if explosive concentrations of material are present. Ensure compliance</li> </ul>
	with applicable exposure limits.
Personal protective equipment Hands: Eyes: Skin and Body: Respiratory:	Wear appropriate protective gloves. Wear appropriate safety glasses. Wear appropriate protective clothes. Wear air-purifying respirator with a tight-fitting facepiece and a high-efficiency particular filter.
[9. Physical and chemical properties ]	
Appearance: Odor: Odor threshold: pH: Melting point and freezing point: Initial boiling point and boiling range: Flash point (Open cup): Evaporation rate (Butyl acetate =1): Flammability (solids, gas): Explosive limits:	White solid (powder) No data available No data available 120 - 140°C No data available No data available No data available No data available No data available No data available

Vapor pressure: Vapor density (Air =1): Specific gravity or density: Solubility: Partition coefficient: n-octanol/water: Auto-ignition temperature: Decomposition temperature: Viscosity:

#### [10. Stability and reactivity]

Reactivity: Chemical stability: Conditions to avoid: No data available No data available Insoluble in water No data available No data available No data available No data available

Not in particular. Stable in general condition. Avoid contact with incompatible materials. Avoid heat, flames, sparks and other sources of ignition. If dry, it can be charged electrostatically by swirling, pneumatic transport, pouring, etc.

Oxidizing materials Oxides of calcium, magnesium and zinc

Hazardous This product will form hazardous fume of oxides of calcium, magnesium and zinc on heating or burning.

Incompatible materials: Hazardous decomposition product: Possibility of hazardous reactions: Heating or combustion reaction:

#### [11. Toxicological information ]

Acute toxicity (Oral):

Acute toxicity (Dermal): Acute toxicity (Gases): Acute toxicity (Vapors): Acute toxicity (Dusts and mists):

Skin corrosion/irritation:

Serious eye damage / Eye irritation:

**Respiratory sensitization:** Skin sensitization: Zinc stearate Based on EC (2008), Oral Rat LD50 is larger than 5000mg/kg. Magnesium stearate Based on IUCLID (2000), Oral Rat LD50 is larger than 10000mg/kg. No data available Not applicable Not applicable Zinc stearate Based on EC (2008), Dermal Rat LC50 is larger than Zinc stearate Based on EC (2008), Dermal Rat LC50 is larger than Magnesium stearate Based on NITE (2006), there is a possibility of skin irritation to human. Zinc stearate Based on NITE (2006), there is a possibility of eye irritation to human. Magnesium stearate Based on NITE (2006), there is a possibility of eye irritation to human. No data available Zinc stearate Based on EC (2008), EU-Risk Assesment Report concluded that "zinc distearate is no like to be skin sensitising", therefore classified as "Not classified".

Germ cell mutagenicity:	Zinc stearate Based on EC (2008), zinc distearate does not have genetic toxicity <i>in vivo</i> and <i>in vitro</i> , therefore classified as "Not classified".
Carcinogenicity:	No data available
Reproductive toxicity:	No data available
STOT-single exposure:	Zinc stearate
	Based on NITE (2006), zinc stearate has a possibility of respiratory irritation to human, therefore classified as "Category 3".
STOT-repeated exposure:	No data available
Aspiration hazard:	No data available
Component analysis - LD50/LC50:	Zinc stearate
	Oral LD50 Rat > 5000mg/kg
	Dermal LD50 Rabbit > 2000mg/kg
	Magnesium stearate
	Oral LD50 Rat > 10000mg/kg
Inhalation (Dust)	
Acute exposure:	Zinc stearate
	May cause irritation, coughing and difficulty breathing. Inhalation of fine dust has produced pneumonia and death in infants. Other effects reported in insufflated dogs included loss of appetite, fever with pneumonitis, peribronchitis and reduction in alveolar size. Magnesium stearate Inhalation of dust may cause mild irritation. Excessive amounts may produce coughing and difficult
Chuonia amaguna	breathing. Zinc stearate
Chronic exposure :	A single case if chronic pneumoconiosis and subsequent fatal lung disease has been reported in a worker exposed for 29 years to zinc stearate. Symptoms included gradually increasing dyspnea and productive cough. Additional effects reported from prolonged inhalation of large amounts include cyanosis, progressive chemical pneumonitis, emphysema, pulmonary edema, and pulmonary granulomatosis.
Skin contact	
Acute exposure:	<ul> <li>Zinc stearate</li> <li>Contact may cause redness. No irritation on intact or broken skin was noted in rats after 24 hours.</li> <li>Magnesium stearate</li> <li>High concentrations may cause unpleasant deposits on the skin. Injury may cause occur by chemical or mechanical action or by the rigorous skin cleansing procedures necessary for removal of the dust.</li> </ul>
Chronic exposure :	Zinc stearate A papular, pustular eczema due to blockage of the sebaceous glands has been reported in workers packing zinc stearate. Irritant granulomas and a single case of contact dermatitis have also been reported.

Eye contact	
Acute exposure :	Zinc stearate Contact may cause irritation with redness and pain.
	Magnesium stearate
	High concentrations may cause unpleasant deposits in
	the eyes, mechanical irritation and seriously reduce
	visibility.
Chronic exposure:	No data available
Ingestion	
Acute exposure:	Zinc stearate
	Large dose may cause abdominal spasms and diarrhea.
	Magnesium stearate
	Ingestion of a large amount of magnesium salts may
	cause diarrhea and abdominal pain.
	More serious symptoms of hypermagnesemia, such as
	electrolyte imbalance, central nervous system depression
	and neurological and cardiac impairment, are rare in the
	absence of intestinal or renal disease since magnesium is
	poorly absorbed from the gastrointestinal tract and
	readily excreted by the kidneys.
Chronic exposure:	Magnesium stearate
	Magnesium preparations may cause phosphorus
	depletion syndrome.

#### [12. Ecological information]

Ecotoxicity	
Aquatic ecotoxicity:	No data available
Terrestrial ecotoxicity:	No data available
Persistence and degradability:	No data available
Bioaccumulative potential:	No data available
Mobility in soil:	No data available
Hazardous to the ozone layer:	No data available

### [13. Disposal considerations]

Do NOT dump this product in the environment or in the household waste. Before disposal or incineration, contents of this product should be neutralized or stabilized if it's possible.

Obey local/regional/national/international regulations about the disposal or the incineration of this product (both contents and containers).

### **[14. Transport information ]**

Not Applicable on UN classification 2915.70 No classification assigned No classification assigned No classification assigned No classification assigned No classification assigned No classification assigned
No classification assigned Not applicable

Particular safety measures for transportation:

Avoid damage to the container while loading this product. Do not put heavy objects on top of this product. Load carefully to prevent the collapse of cargo. Avoid direct sunlight to this product during transport.

### [15. Regulatory information ]

#### **Inventory information**

Inventory Name	Zinc stearate		Magnesium stearate	
	Status	Registry Number	Status	Registry Number
AICS (Australia):	Present	_	Present	—
DSL (Canada):	Present	_	Present	_
IECSC (China):	Present	30048	Present	30034
EINECS (EU):	Present	209-151-9	Present	209-150-3
ENCS (Japan):	Present	(2)-615	Present	(2)-611
KECL (Korea):	Present	KE-26418	Present	KE-26390
INSQ (Mexico):	Present	_	Present	_
NZIoC (New Zealand):	Present	HSR003105	Present	_
PICCS (Philippines):	Present	_	Present	_
TCSI (Taiwan):	Present	_	Present	_
Inventory (Turkey):	Present	EC No. 209-151-9	Present	EC No. 209-150-3
TSCA (U.S.A.):	Present	_	Present	_

#### [16. Other Information ]

#### References

1	ChemADVISOR, Inc. (2014). Magnesium stearate [Data file]. Retrieved from LOLI database.
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3	European Communities. (2008). Risk Assessment Report. Zinc Distearate (Final Report), 44.
4	IUCLID. (2000). Dataset for Magnesium stearate [Data file].
5	IUCLID. (2000). Dataset for Zinc distearate [Data file].
6	Japan Chemical Database Ltd. (2015). <i>Magnesium stearate</i> [Data file]. Retrieved from ezADVANCE database.
7	Japan Chemical Database Ltd. (2015). Zinc stearate [Data file]. Retrieved from ezADVANCE database.

8	National Institute of Technology and Evaluation (NITE). (2006). Classification resulte(ID801-900 [Data file]. Retrieved from						
		http://www.safe.nite.go.jp/english/files/ghs_xls/classification_result_e(ID801-900).xls.					
9	National Institute of Technology and Evaluation (NITE). (2006). <i>Magnesium stearate</i> [Data file].						
10							
11	United Nations. (2013). <i>Globally Harmonized System of Classification and Labelling of Chemicals</i> ( <i>GHS</i> ) (5th ed.). (The Japanese GHS Inter-ministerial Committee, Trans.). Tokyo: The Chemical Daily Co., Ltd						
Key/ Legend							
ACGIH - Am	erican Conference of Governmen	tal Industrial Hygienists					
AICS - Austra	alia Inventory of Chemical Substa	ances					
ADR - European Road Transport		CAS - Chemical Abstracts Service					
°C - degree C	elsius	DSL - Domestic Substances List					
EINECS - Eu	ropean Inventory of Existing Cor	nmercial Chemical Substances (European Union)					
ENCS - Exist	ing and New Chemical Substance	es (Japan)					
GHS - Global	lly Harmonized System of Classif	fication and Labelling of Chemicals					
HPV - High F	Production Volume	HS code - Harmonized System code					
IATA - Intern	national Air Transport Association	Ω					
ICAO - Interr	national Civil Aviation Organizati	on					
IECSC - Inve	ntory of Existing Chemical Subst	ances (China)					
IMDG - Inter	national Maritime Dangerous Go	ods					
INSQ - Natio	nal Inventory of Chemical Substa	inces (Mexico)					
IUCLID - Inte	ernational Uniform Chemical Info	ormation Database					
KECL - Kore	a Existing Chemicals Inventory	NITE - National Institute of Technology and Evaluation					
LD50 - Letha	l Dose, 50% or Median Lethal De	ose					
LOLI - List C	Of Lists <sup>TM</sup> -ChemADVISOR's Reg	ulatory Database					
NZIoC - New	Zealand Inventory of Chemicals						
PICCS - Phili	ppines Inventory of Chemicals an	nd Chemical Substances					
	istant of Tania Effects of Chamies						

PICCS - Philippines Inventory of Chemicals and Chemica	al Substances
RTECS - Registry of Toxic Effects of Chemical Substance	RID - European Rail Transport
STOT - Specific Target Organ Toxicity	TCSI - Taiwan Chemical Substance Inventory
TDG - Transportation of Dangerous Goods	TLV - Threshold Limit Value
TSCA - Toxic Substances Control Act (U.S.A.)	TWA - Time Weighted Average
UN - United Nations	US DOT - United States Department of Transportation

#### Manufacture disclaimer

All information given in this SDS is based on the data which is considered to be accurate, but the information do not guarantee enough safety. All chemical material may have an unknown hazard to human and conditions of methods of handling, storage, use and disposal of the product are beyond suppliers' control; therefore all risks and consequences of use the product are on users' responsibilities and users need to set appropriate safety measures for special use.

In addition, all classification in this SDS was written in accordance with the GHS classification of the fifth revised edition. However, GHS mentioned that countries are free to determine which of the building blocks will be applied in different parts of label elements and building blocks. Therefore, many countries set own requirements of label elements and building blocks. In the cases of export from Japan or use in other countries, SDSs and labels are needed, which are in accordance with the local laws and regulations of exporting countries or user countries. Please contact supplier beforehand for checking SDSs and labels are suitable for the local laws and regulations.